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Microbial transformation of dehydrocostuslactone by *Mucor polymorphosporus*

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Biotransformation of dehydrocostuslactone (**1**) by *Mucor polymorphosporus* yielded four compounds, and their structures were identified as 11 α ,13-dihydrodehydrocostuslactone (**2**), 3 α -hydroxy-11 α ,13-dihydrodehydrocostuslactone (**3**), 3 β -hydroxy-4 β ,15,11 α ,13-tetrahydrodehydrocostuslactone (**4**) and 2 β -hydroxy-11 α ,13-dihydrodehydrocostuslactone (**5**), respectively, on the basis of their spectral data. Among them, compound 5 is a new compound.

Keywords: Biotransformation; Dehydrocostuslactone; *Mucor polymorphosporus*; *Aucklandia lappa*

1. Introduction

Aucklandia lappa is a commonly used medicinal plant and its rhizome has been used as a traditional Chinese medicine possessing anti-cancer, anti-oxidation and anti-fungus activities [1,2]. Dehydrocostuslactone (**1**), with a guaiane skeleton, is one of the major active constituents in this medicinal herb. Recently, Lee *et al.* reported that dehydrocostuslactone inhibited the expression of inducible nitric oxide (NO) synthases and tumour necrosis factor- α in lipopolysaccharide-activated macrophages [3]. Biotransformation is a useful tool to modify structures of biologically active compounds. In recent years, applying microbial transformation for studying the metabolism of natural products has been approached [4,5]. We have also fulfilled a series of biotransformation studies to optimise the structures of bioactive natural products and to pursue the mammalian metabolism by a microbial approach [6–16]. In this paper, we report the biotransformation of **1** by filamentous fungi, with the aim of improving its solubility and activities.

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2. Results and discussion

Incubation of dehydrocostuslactone with *Mucor polymorphosporus* for 4 days yielded four products (figure 1), their structures being identified as 11 α ,13-dihydrodehydrocostuslactone (**2**), 3 α -hydroxy-11 α ,13-dihydrodehydrocostuslactone (**3**), 3 β -hydroxy-4 β ,15,11 α ,13-tetrahydrodehydrocostuslactone (**4**) and 2 β -hydroxy-11 α ,13-dihydro-dehydrocostuslactone (**5**), respectively, on the basis of their spectral data. Among them, **5** is a new compound.

According to TLC analysis, the biotransformed products were more polar than the substrate. Site-specific hydroxylation and hydrogenation reaction of the substrate were observed in the biotransformation process of **1**.

EI-MS of compound **5** showed a molecular ion peak at m/z 248 $[M]^+$, together with its ^1H NMR and ^{13}C NMR data, indicating the molecular formula of $\text{C}_{15}\text{H}_{20}\text{O}_3$. The methylene signals of C-13 were also replaced by a methyl doublet. The NOE enhancement between H-13 and H-6 suggested that the relative configuration of 13-Me should be in β -configuration. In comparison with that of compound **2**, the ^{13}C NMR spectrum of **5** exhibited an additional oxygen-bearing methine signal at δ 73.8, and the HSQC spectrum exhibited the disappearance of a methylene, suggesting the introduction of a hydroxyl group at methylene in the molecule. In the HMBC spectrum, H-2 (δ 4.09) correlated with C-3 (δ 42.4), C-5 (δ 49.8), C-1 (δ 55.4), C-10 (δ 147.8) and C-4 (δ 151.7); meanwhile H₂-15 (δ 5.06 and 4.93) correlated with carbon signals at C-3 (δ 42.4) and C-5 (δ 49.8), suggesting the hydroxyl group should be at C-2 position (see figure 2). The NOE

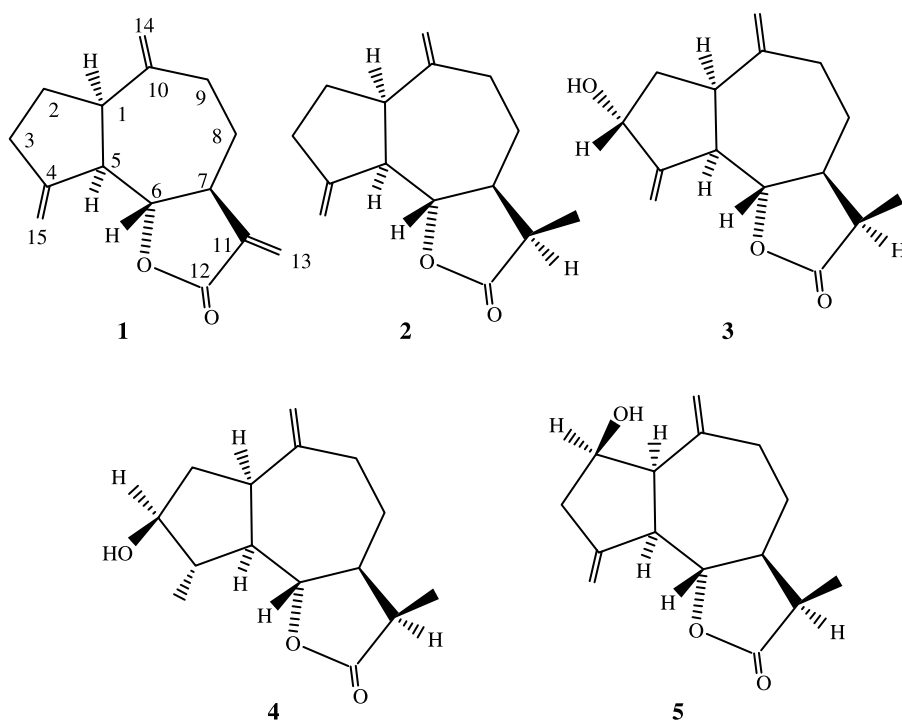


Figure 1. Chemical structures of compounds **1**–**5**.

Table 1. ^1H NMR data of compounds **2**, **3**, **4** and **5** (in $\text{DMSO}-d_6$, 500 MHz).

^1H	2	3	4	5
1	2.90 (1H, m)	2.96 (1H, m)	2.65 (1H, m)	2.67 (1H, m)
2 α	1.82 (1H, m)	1.67 (1H, m)	1.54 (1H, m)	4.09 (1H, m)
2 β	1.90 (1H, m)	2.00 (1H, m)	1.92 (1H, m)	–
3 α	2.53 (1H, m)	–	3.46 (1H, m)	2.32 (1H, m)
3 β	2.53 (1H, m)	4.46 (1H, br.s)	–	2.78 (1H, m)
4	–	–	1.64 (1H, m)	–
5	2.79 (1H, m)	2.94 (1H, m)	1.83 (1H, m)	2.94 (1H, m)
6	4.01 (1H, t, $J = 9.5$ Hz)	3.93 (1H, t, $J = 9.5$ Hz)	3.93 (1H, t, $J = 9.5$ Hz)	3.92 (1H, t, $J = 9.5$ Hz)
7	2.42 (1H, m)	2.48 (1H, m)	2.24 (1H, m)	2.00 (1H, m)
8 α	1.39 (1H, m)	1.23 (1H, m)	1.21 (1H, m)	1.27 (1H, m)
8 β	1.76 (1H, m)	1.81 (1H, m)	1.81 (1H, m)	2.05 (1H, m)
9 α	1.99 (1H, m)	1.94 (1H, m)	1.80 (1H, m)	2.02 (1H, m)
9 β	2.42 (1H, m)	2.43 (1H, m)	2.48 (1H, m)	2.41 (1H, m)
11	2.62 (1H, dq, $J = 8.0, 8.0$ Hz)	2.61 (1H, dq, $J = 8.0, 8.0$ Hz)	2.61 (1H, dq, $J = 8.0, 8.0$ Hz)	2.62 (1H, dq, $J = 8.0, 8.0$ Hz)
13	1.05 (1H, d, $J = 7.5$ Hz)	1.03 (3H, d, $J = 7.5$ Hz)	1.01 (3H, d, $J = 7.5$ Hz)	1.09 (3H, d, $J = 7.5$ Hz)
14a	4.84 (1H, br.s)	4.85 (1H, br.s)	4.86 (1H, br.s)	4.53 (1H, br.s)
14b	4.71 (1H, br.s)	4.67 (1H, br.s)	4.83 (1H, br.s)	4.84 (1H, br.s)
15a	5.04 (1H, br.s)	5.15 (1H, br.s)	1.07 (3H, d, $J = 7.5$ Hz)	5.06 (1H, br.s)
15b	4.97 (1H, br.s)	5.12 (1H, br.s)	–	4.93 (1H, br.s)
OH	–	4.97 (br.s)	4.77 (br.s)	4.53 (br.s)

3.3 Culture medium

All culture and biotransformation experiments using filamentous fungi were performed in potato medium, which was made by the following composition (l): 200 g potato and 20 g glucose.

3.4 Screening test

Twenty strains of filamentous fungi were screened for their capabilities to transform **1**, which were carried out in 250-ml Erlenmeyer flasks containing 100 ml of the liquid culture. The flasks were placed on rotary shakers, operating at 180 rpm at 26–28°C. The substrates were dissolved in acetone to reach a concentration of 10 mg/ml. After 36 h of pre-culture, 0.2 ml of the solution was added into each flask and these flasks were maintained under the fermentation condition for 4 days. Culture controls consisted of fermentation blanks in which micro-organisms were grown without substrate but with the same amount of acetone. Substrate controls contained the sterile medium with the same amount of substrate and incubated under the above conditions.

3.5 Preparative biotransformation by *M. polymorphosporus*

Preparative scale biotransformation of dehydrocostuslactone by *M. polymorphosporus* was carried out in a 1000-ml Erlenmeyer flask. The substrate (10 mg) in 1 ml acetone was added to 350 ml pre-cultured medium for 36 h. In total, 400 mg of substrate was used. The incubation was continued for four additional days. Other procedures were the same as for screening scale biotransformation.

The culture was filtered and the filtrate extracted with same volume of EtOAc for five times. The organic phase was collected and concentrated to dryness in vacuo. The residues were applied to a silica gel column and eluted with petroleum ether/ethyl acetate (in a gradient manner from 100:3 [v/v] to 1:1, at a flow rate of 1.5 ml/min) to obtain compounds **1** (61 mg), **2** (50 mg), **3** (13 mg), **4** (15 mg) and **5** (4 mg), respectively. All of the products were identified on the basis of their spectral data.

Table 2. ^{13}C NMR chemical shifts of the compounds **2**, **3**, **4** and **5** (in DMSO- d_6 , 125 MHz).

Carbons	2	3	4	5
1	46.3	42.6	41.4	55.4
2	28.0	39.5	38.2	73.8
3	32.0	72.6	76.6	42.4
4	150.3	155.6	46.4	151.7
5	51.9	49.6	50.1	49.8
6	84.5	84.9	85.7	85.2
7	43.3	43.6	46.4	47.9
8	29.4	28.2	28.6	31.2
9	36.9	37.5	36.3	38.1
10	152.4	150.1	150.1	147.8
11	38.5	38.5	38.6	40.2
12	179.4	179.5	179.3	178.3
13	11.0	11.0	10.9	12.8
14	111.2	110.1	111.5	111.6
15	108.1	111.4	18.2	108.3

3.6 Structural identification

3.6.1 Compound 2. Colourless oil (Et₂O). $[\alpha]_D^{22} + 90$ (*c* 0.6, CHCl₃). UV λ_{\max} (MeOH): 210. ¹H NMR and ¹³C NMR data: see tables 1 and 2.

3.6.2 Compound 3. Colourless oil (Et₂O). $[\alpha]_D^{22} + 28.2$ (*c* 2.0, CHCl₃). UV λ_{\max} (MeOH): 210. ¹H NMR and ¹³C NMR data: see tables 1 and 2.

3.6.3 Compound 4. Colourless oil (Et₂O). $[\alpha]_D^{22} + 35.2$ (*c* 1.8, CHCl₃). UV λ_{\max} (MeOH): 210. IR (KBr) ν_{\max} (cm⁻¹): 3400, 1758, 1620, 870, 810. ¹H NMR and ¹³C NMR data: see tables 1 and 2. EI-MS (*m/z*): 250 [M]⁺, 105, 91, 80. HRFAB-MS (*m/z*): 249.1503 [M - H]⁺ (calcd for C₁₅H₂₁O₃, 249.1489).

3.6.4 Compound 5. Colourless oil (Et₂O). $[\alpha]_D^{22} + 21.2$ (*c* 2.0, CHCl₃). UV λ_{\max} (MeOH): 210. IR (KBr) ν_{\max} (cm⁻¹): 3460, 1720, 1610, 890, 810. ¹H NMR and ¹³C NMR data: see tables 1 and 2. EI-MS (*m/z*): 248 [M]⁺, 174, 105, 91, 55. HRFAB-MS (*m/z*): 247.1341 [M - H]⁺ (calcd for C₁₅H₁₉O₃, 247.1334).

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